



Reaction of vinylcarbenoids with thioketones: formation of vinylthiocarbonyl ylides followed by ring closure to thiiranes and dihydrothiophenes

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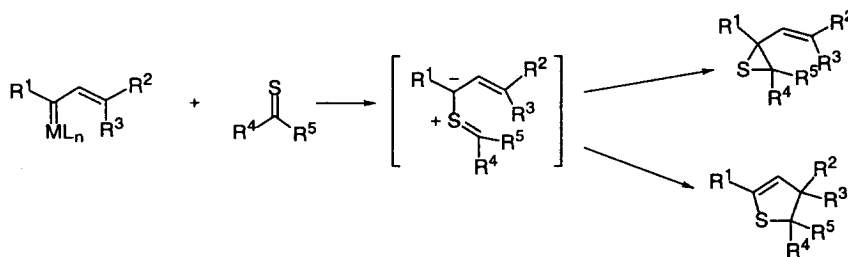
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Abstract

Reactions were carried out on thioketones with vinylcarbenoids. Rh₂(OAc)₄-catalyzed reaction of vinyldiazo compounds **1a,b** with xanthione **2** and thiochromone **3** gave thiiranes **4b** and olefins **5** and **6**, desulfurization products from thiiranes, while the reaction of **1c** with **2** and **3** afforded dihydrothiophenes **7**, **8** and **9**. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: carbenes; carbenoids; thiocarbonyl compounds; cyclization; ylides.

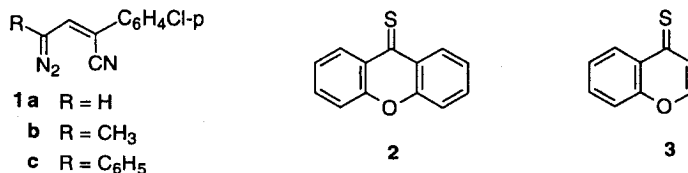
Reaction of diazoalkanes with thioketones giving unstable 1,3,4-thiadiazolines followed by extrusion of nitrogen has been widely studied by Schönberg et al. as a traditional method of generation of thiocarbonyl ylides.^{1–4} The reaction of metallocarbenoids, generated from diazo compounds with thiocarbonyl compounds such as thioamides and thionesters, has recently attracted much interest from a synthetic viewpoint (Scheme 1).⁵ From a synthetic and mechanistic viewpoint the chemistry of vinylcarbenoids has received significant attention over the past ten years.⁶ Davies et al. have found that the vinylic β-carbon of vinylcarbenoids is involved in the reaction with dienes, giving [3+4] cycloadducts via formation of *cis*-divinylcyclopropanes followed by Cope rearrangement.⁷ We have also expected the participation of the vinylic β-carbon of vinylcarbenoids in the reaction with heteroatom compounds.⁸



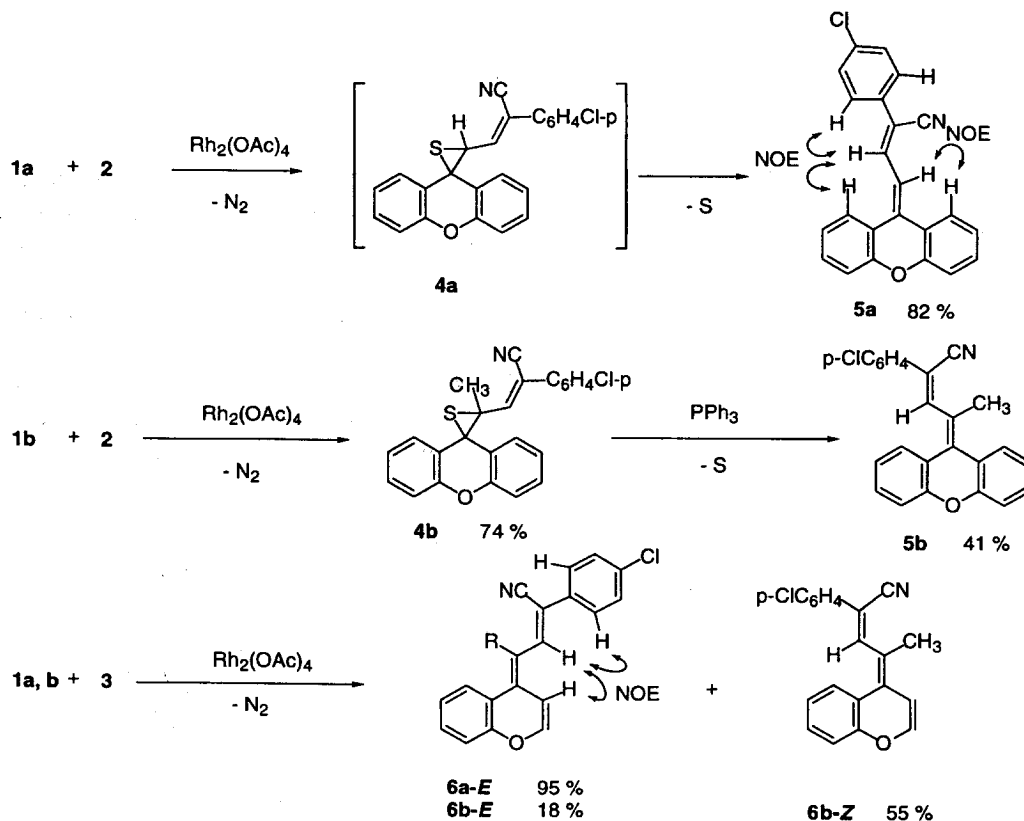
Scheme 1.

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In the present communication, we report the first example of generation of novel vinylthiocarbonyl ylides by the reaction of vinylcarbenoids with thioketones. Metallovinylcarbenoids generated from $\text{Rh}_2(\text{OAc})_4$ -catalyzed decomposition of vinyldiazo compounds attack a lone pair of sulfur of thioketones to give vinylthiocarbonyl ylides, which could undergo 1,3- and 1,5-electrocyclization to thiiranes and dihydrothiophenes, respectively. We used vinyldiazo compounds **1a–c** as precursors of carbenoids, and xanthione **2** and thiochromone **3** as thiocarbonyl compounds.

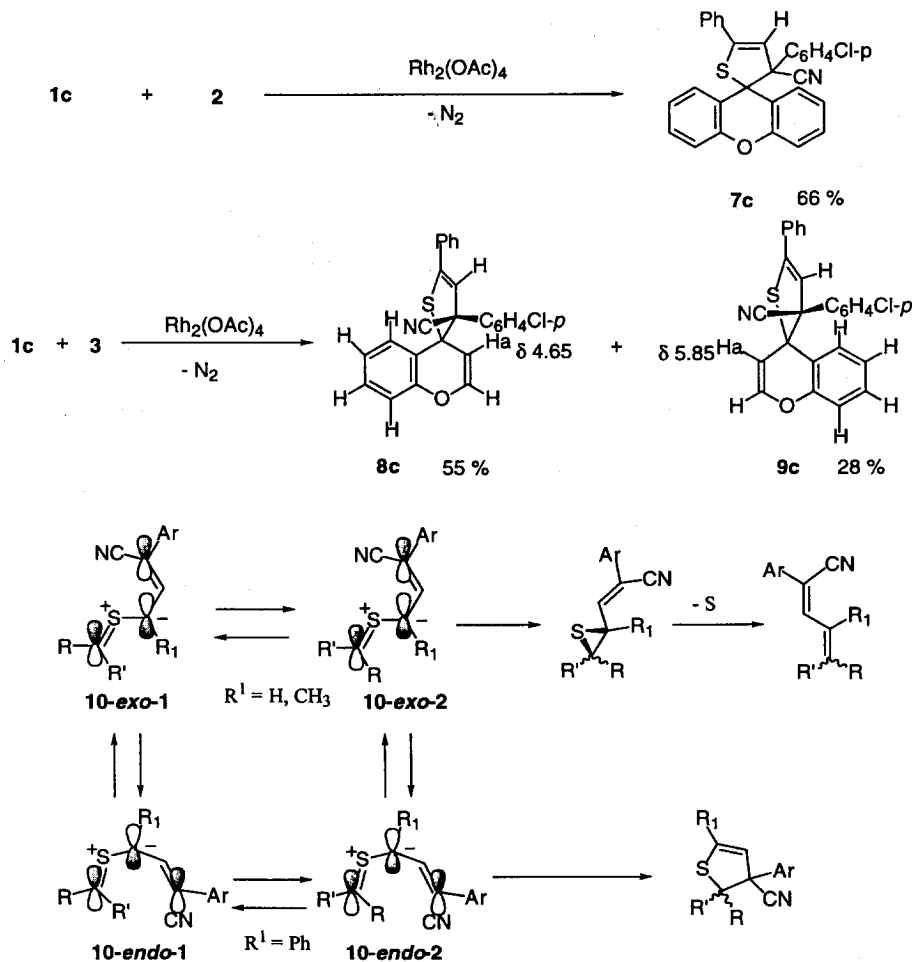


2-(*p*-Chlorophenyl)-4-diazo-*cis*-2-butenenitrile (**1a**) in dichloromethane was added over 6 hours to a solution of an equimolar amount of xanthione **2** and a catalytic amount of $\text{Rh}_2(\text{OAc})_4$ at room temperature and the reaction mixture was separated by chromatography over silica gel to give the diene **5a** in the yield of 82% (Scheme 2). The thiirane **4b** was isolated in the yield of 74% in the reaction of 2-(*p*-chlorophenyl)-4-diazo-*cis*-2-petenenitrile (**1b**) with **2** in the presence of $\text{Rh}_2(\text{OAc})_4$. Treatment of **4b** with triphenylphosphine in refluxing THF gave the diene **5b**. The reaction of **1a** with thiochromone **3** afforded the diene **6a-E** in the yield 95%, while **1b** reacted with **3** to yield two isomeric dienes **6b-E** and **6b-Z** in the yields of 18 and 55%, respectively.⁹ Stereochemistry of the dienes **6a-E**, **6b-E** and **6b-Z** was determined by NOE experiments. Thus, both **1a** and **1b** gave dienes or a thiirane in the reaction with xanthione **2** and thiochromone **3**.



Scheme 2.

However, phenylvinyl diazomethane derivative **1c** reacted with **2** to give only dihydrothiophene **7c** (66%) (Scheme 3). Two isomeric spirocyclic dihydrothiophene **8c** and **9c** were obtained in a ratio of 2:1 in the reaction with **3**. The structures of **8c** and **9c** were assigned on the basis of the appearance of an Ha proton (δ 4.65) of **8c** at 1.20 ppm higher field due to the shielding effect of chlorophenyl group *syn* to Ha comparing with that of **9c** (δ 5.85).¹⁰



Scheme 3.

Vinylcarbenoids attack either of the lone pairs of sulfur of the thioketones to give vinylthio-carbonyl ylides. In the reaction of **1a** with **2**, only sterically favorable vinylthiocarbonyl ylide **10-exo** ($R=R'$), in which a cyanostyryl group is located at *exo*-position, could be formed and cyclized in a conrotatory fashion to the thiirane **4a** followed by extrusion of sulfur, yielding the diene **5a**. The reaction of **1a** with **3** would produce two isomeric vinylthiocarbonyl ylides **10-exo-1** and **10-exo-2**, which would undergo cyclization to isomeric thiiranes and subsequent desulfurization would give **6b-E** and **6b-Z**. The vinylcarbenoid from **1c** attacks either of sulfur lone pairs of **3** to give two isomeric vinylthiocarbonyl ylides **10-endo-1** and **10-endo-2**, in which a larger phenyl group is located at the *exo*-position and a smaller cyanostyryl group at the *endo*-position. The ylides **10-endo-1** and **10-endo-2** could undergo 6-electrons cyclization to dihydrothiophene derivatives **8c** and **9c**.

In conclusion, novel vinylthiocarbonyl ylides, generated from the reaction of vinylcarbenoids with thioketones, undergo 1,3-electrocyclization to a thiirane or 1,5-electrocyclization to a dihydrothiophene depending on the properties of substituents of the ylides.

References

1. Schönberg, A.; König, B.; Singer, E. *Chem. Ber.* **1967**, *100*, 767.
2. (a) Huisgen, R.; Langhals, E. *Tetrahedron Lett.* **1989**, *30*, 5369. (b) Huisgen, R.; Xingya, L. *Tetrahedron Lett.* **1983**, *24*, 4185. (c) Huisgen, R.; Xingya, L. *Heterocycle* **1983**, *20*, 2363. (d) Kalwisch, I.; Xingya, L.; Gottstein, J.; Huisgen, R. *J. Am. Chem. Soc.* **1981**, *103*, 7032.
3. Mataka, S.; Ishi-i, S.; Tashiro, M. *J. Org. Chem.* **1978**, *43*, 3730.
4. (a) Kägi, M.; Linden, A.; Mloston, G.; Heimgartner, H. *Helv. Chim. Acta.* **1998**, *81*, 285. (b) Mloston, G.; Heimgartner, H. *Helv. Chim. Acta.* **1996**, *79*, 1785. (c) Kägi, M.; Linden, A.; Mloston, G.; Heimgartner, H. *Helv. Chim. Acta.* **1996**, *79*, 855. (d) Mloston, G.; Petit, M.; Linden, A.; Heimgartner, H. *Helv. Chim. Acta.* **1994**, *77*, 435.
5. (a) Takano, S.; Tomita, S.; Takahashi, M.; Ogasawara, K. *Synthesis* **1987**, 1116. (b) Fang, F. G.; Prato, M.; Kim, G.; Danishefsky, S. J. *Tetrahedron Lett.* **1989**, *30*, 3625. (c) Fang, F. G.; Danishefsky, S. J. *Tetrahedron Lett.* **1989**, *30*, 2747. (d) Fang, F. G.; Maier, M. E.; Danishefsky, S. J. *J. Org. Chem.* **1990**, *55*, 831. (e) Kim, G.; Chu-Moyer, M. Y.; Danishefsky, S. J. *J. Am. Chem. Soc.* **1990**, *112*, 2003. (f) Kim, G.; Chu-Moyer, M. Y.; Danishefsky, S. J.; Shulte, G. K. *J. Am. Chem. Soc.* **1993**, *115*, 30. (g) Mlostón, G.; Heimgartner, H. *Helv. Chim. Acta.* **1996**, *79*, 1785.
6. (a) Gu, H. H.; McDaniel, K. F.; McMills, M. C.; Yap, G. P.; Rheingold, A. L. *Tetrahedron Lett.* **1997**, *38*, 6993. (b) Davies, H. M. L.; Bruzinski, P. R.; Lake, D. H.; Kong, N.; Fall, M. J. *J. Am. Chem. Soc.* **1996**, *118*, 6897. (c) Davies, H. M. L.; Doan, B. D. *Tetrahedron Lett.* **1996**, *37*, 3967. (d) Davies, H. M. L.; Hansen, T.; Rutberg, J.; Bruzinski, P. R. *Tetrahedron Lett.* **1997**, *38*, 1741. (e) Yoshikoshi, K.; Achiwa, K. *Chem. Pharm. Bull.* **1995**, *43*, 2048.
7. (a) Davies, H. M. L. *Tetrahedron* **1993**, *49*, 5203. (b) Davies, H. M. L.; Doan, B. D. *J. Org. Chem.* **1998**, *63*, 657. (c) Davies, H. M. L.; Clark, T. J. *Tetrahedron* **1994**, *50*, 9883.
8. Hamaguchi, M.; Misumi, T.; Oshima, T. *Tetrahedron Lett.* **1998**, *39*, 7113.
9. As the *trans* structure of α -cyanostyryl group was confirmed by NOE experiments, the suffixes *E*, *Z* are used to denote the stereochemistry of another double bond of the dienes.
10. ^1H NMR (CDCl_3) data for **8c**: δ 4.65 (1H, J=6.3 Hz), 5.99 (s, 1H), 6.36 (d, 1H, J=6.3 Hz), 6.97 (dd, 1H, J=8.2, 1.4 Hz), 7.18–7.24 (m, 1H), 7.25–7.39 (m, 5H), 7.44–7.48 (m, 2H), 8.28 (dd, 1H, J=7.9, 1.6 Hz). ^1H NMR (CDCl_3) data for **9c**: δ 5.85 (d, 1H, J=6.3 Hz), 5.96 (s, 1H), 6.68 (dd, 1H, J=8.5, 1.4 Hz), 6.80–6.86 (m, 1H), 6.89 (d, 1H, J=6.3), 7.05–7.14 (m, 5H), 7.30 (dd, 1H, J=7.9, 1.3 Hz), 7.45–7.48 (m, 3H), 7.62–7.66 (m, 2H).